

Journal of Alloys and Compounds 320 (2001) 18–23

Several of
ALLOYS
AND COMPOUNDS
————————————————————

www.elsevier.com/locate/jallcom

Ferromagnetism in the strontium deficient $La_{0.67}Sr_{0.15}\Box_{0.18}MnO_3$ perovskite manganite

L. Laroussi^a, A. Cheikhrouhou^{a, *}, J. Pierre^b

a *Laboratoire de Physique des Materiaux ´ ´* , *Faculte des Sciences de Sfax*, *B*. *P*. 802, ³⁰¹⁸ *Sfax*, *Tunisia* b *Laboratoire Louis NEEL*, *CNRS*, *B*. *P*. ¹⁶⁶*X*, ³⁸⁰⁴² *Grenoble*, *France*

Received 22 December 2000; accepted 16 January 2001

Abstract

The structural, magnetic and magnetotransport properties of Sr deficient perovskite manganite oxide $La_{0.67}Sr_{0.15}\square_{0.18}MnO_3$ have been investigated. X-ray diffraction patterns have been indexed in the rhombohedral str showed that our sample exhibits a paramagnetic–ferromagnetic transition at T_c = 360 K, while electrical measurements show a maximum resistivity at $T_o = 210$ K. Strontium deficiency led to a strong decrease of the electrical transition temperature. At low temperature (20 K), a magnetic applied field of about 0.3 T induced a decrease of the resistivity of about 85%. At low temperature, our sample exhibited a magnetoresistance effect of about 30% under a magnetic applied field of about 4 T. \circ 2001 Elsevier Science B.V. All rights reserved.

Keywords: Ferromagnetism; Strontium deficient perovskite manganite

With the discovery of the very large negative magneto-

Teller distortion [22]. resistance, great interest has been focused on the perov- Many studies are being carried out on the magnetic and skite-type hole doped manganese oxides with the general magneto-transport properties of these materials. Effects of formula $Ln_{1-x}M_xMnO_3$ where Ln is a trivalent rare-earth divalent alkaline-earth elements have been extensively and M is a divalent alkali-earth [1-8]. These systems have studied [7-14]. These studies showed that the Cur been given much attention since they could possibly be
used for a Mn⁴⁺ content of about 33%. However,
used for sensor applications [9-11]. Different substitutions optimized for a Mn⁴⁺ content of about 33%. However, in these systems lead to different crystal structures, spin only few studies have been carried out on deficiency states and transport properties. The replacement of the effects in manganite systems [23–25]. Deficiencies in the trivalent element by a divalent or monovalent element A perovskite site lead to an increase in the Mn⁴⁺ c charge neutrality and subsequently a modification occurs $La_{0.67}Sr_{0.33}MnO_3$, we investigate the structural, magnetic of both magnetic and transport properties. Such modi-
and electrical properties in the $La_{0.67}Sr_{0.15$ of both magnetic and transport properties. Such modi-
fications can be understood on the basis of double-ex-
deficient compound. In such sample, vacancy implies an change interactions between the spins of the Mn³⁺ and increase of the Mn⁴⁺ content beyond 33% and also a Mn⁴⁺ ions [12,13]. However, recent studies have shown change in the average ionic radius $\langle r_A \rangle$ of the A-sit that the double exchange alone cannot explain the observed behaviors in these systems and suggested, that other effects ruled by average A-site cationic radius $\langle r_A \rangle$ [14– **2. Experimental** 16], A-site cationic size mismatch [17,18], and oxygen

1. Introduction deficiency $[19-21]$ play a crucial role as to Mn valence, charge ordering or pinning, polaron effects and Jahn–

studied $[7-14]$. These studies showed that the Curie

A powder sample of $La_{0.67}Sr_{0.15}\square_{0.18}MnO_3$ was prepared by a conventional solid–solid reaction. The precursors La_2O_3 , SrCO₃ and MnO₂ in the form of high-*Corresponding author. Tel.: $+216-4-274-390$; fax: $+216-4-274-437$. purity (>99%) powders were fired in air at 700°C for 10 h *E*-*mail address*: abdel.cheikhrouhou@fss.rnu.tn (A. Cheikhrouhou). before being used. These materials were thoroughly mixed

in an agate mortar, in stoichiometric proportions and Table 1 heated in air at 900°C for 48 h to achieve decarbonation. The resulting powder was then pressed into pellets form of about 1 mm thickness and fired at 1350° C for 36 h in air with several periods of grinding and sintering. Finally, the pellets were rapidly quenched to room temperature in water. This final step was carried out in order to retain the structure obtained at the sintered temperature 1350°C. In **3. Results and discussion** fact, studies on $Pr_{0.5}Sr_{0.5}MnO_3$ [26,27] showed that the quenching method has an important effect on the physical X-ray analysis shows that our sample exhibits practicalproperties of these manganites. ly identical X-ray powder patterns as stoichiometric

recorded by a laboratory-made Faraday balance, whereas data for both samples. the susceptibility in 0.05 T was recorded with a vibrating In order to check the oxygen stoichiometry, our sample sample magnetometer. Magnetization measurements versus was annealed at 800°C in air for 2 weeks. XRD patterns magnetic applied field up to 8 T were recorded by an using a diffractometer $(\lambda = 1.54056 \text{ Å})$ of both stoichioextraction magnetometer in the temperature range 20–430 metric and deficient annealed samples (Fig. 2) showed no

and the magnetic applied field were performed on dense We plot in Fig. 3 the magnetization evolution versus ceramic pellets by the conventional four-probe technique temperature of both stoichiometric and lacunar samples. in the temperature range 20–300 K. Our sample exhibits a clear transition from a paramagnetic

Phase purity, homogeneity and lattice parameters were $La_{0.67}Sr_{0.33}MnO_3$. We plot in Fig. 1 the X-ray diffraction determined by X-ray powder diffraction using Guinier- (XRD) patterns at room temperature for both stoich (XRD) patterns at room temperature for both stoichio-Hagg cameras, with CrK α_1 radiation, at room temperature. metric $La_{0.67}Sr_{0.33}MnO_3$ and Sr deficient High-purity silicon powder was used as an internal stan-
dard. Unit cell parameters were obtained by least squares are indexed in a perovskite structure with a distortion from are indexed in a perovskite structure with a distortion from calculations.
Magnetization measurements versus temperature were *R3_c* space group. We list in Table 1 the crystallographic

K. Significant change in the structure. This result confirms that Electrical measurements as a function of the temperature our deficient sample is not substoichiometric in oxygen.

Fig. 1. X-ray diffraction patterns for both $La_{0.67}Sr_{0.15}\square_{0.18}MnO_3$ and $La_{0.67}Sr_{0.33}MnO_3$ samples.

Fig. 2. X-ray diffraction patterns for both annealed $\text{La}_{0.67}\text{Sr}_{0.15}\square_{0.18}\text{MnO}_3$ and $\text{La}_{0.67}\text{Sr}_{0.33}\text{MnO}_3$ samples.

 $La_{0.67}Sr_{0.15}\Box_{0.18}MnO_3$ and $La_{0.67}Sr_{0.33}MnO_3$ at $H=500$ Oe. temperatures for $La_{0.67}Sr_{0.15}\Box_{0.18}MnO_3$.

to a ferromagnetic state with decreasing temperature. The netic applied field up to 8 Tesla in the temperature range Curie temperature is found to be 360 K. Strontium 20–430 K. The magnetization exhibits a sharp increase at deficiency induces a slight decrease in T_c , which is found very low magnetic applied field then saturates at 1T. This to be 375 K for the stoichiometric sample. This result confirms the ferromagnetic behavior below 360 result confirms the ferromagnetic behavior below 360 K.

Fig. 4 shows the magnetization evolution versus mag-
ture, we have plotted in Fig. 5 Arrott curves $(M^2 \text{ vs.})$ *H*/*M*). The Curie temperature is found to be 360 K.

Fig. 3. Temperature dependence of the magnetization for Fig. 4. Magnetization evolution versus applied magnetic field at several

spontaneous magnetization $M_{\rm sp}$ at 10 K is about 2.8 $\mu_{\rm B}$ / using the following equation exchange interaction.

$$
\cos\left(\frac{\phi}{2}\right) = \frac{M_{\rm sp}}{M_{\rm sth}} = \frac{2.8}{3.31} = 0.85
$$

Fig. 6. Spontaneous magnetization and reciprocal susceptibility data mum becomes broader. versus temperature for $La_{0.67}Sr_{0.15}\Box_{0.18}MnO_3$. In the semiconductor-like phase, the conduction is

dence of the magnetic moment per mole $M_{\rm sn}(T)$ can be described by the Brillouin–Weiss molecular field theory modeling the degree of spin collinearity.

In the paramagnetic phase, the inverse of the susceptibility versus temperature exhibits the expected Curie–Weiss law $\chi = C/(T - \theta_n)$. From the linearity of the χ^{-1} *T* curve, a net transition between the paramagnetic and the ferromagnetic states is confirmed corresponding to a paramagnetic Curie temperature $\theta_{p} = 363.8$ K. The obtained Curie constant *C* is 0.48 K $\cdot \mu_B / kOe$. For the stoichiometric sample $La_{0.67}Sr_{0.33}MnO_3$, θ_p and *C* are found to be 375 K and 0.65 K $\cdot \mu_B / kOe$, respectively. The strontium deficiency leads to a decrease in the T_c and θ_p values.

As a strontium vacancy induces an increase in the Mn^{4+} content, the T_c decrease observed in our sample can be explained by the increase of the Mn⁴⁺ concentration which ² leads to a decrease of the double exchange interactions. Fig. 5. Arrott curves *^M* (*H*/*M*) for La Sr ^h MnO . 0.67 0.15 0.18 3 This result is in concordance with previous work on $La_{1-x}Sr_xMnO_3$ which shows a parabolic behavior of T_C
Fig. 6 shows the temperature dependence of the sponta- versus Mn^{4+} content with a maximum obtained at about neous magnetization (M_{sp}) and the inverse susceptibility 33% [28]. However, the double exchange mechanism (χ^{-1}) as function of temperature in our sample. The based on the Mn⁴⁺:Mn³⁺ ratio alone cannot explain the $M_{\rm sp}(T)$ curve drops rapidly near $T_{\rm C}$ = 360 K showing a observed behaviors in the diluted manganese oxides. The well defined Curie temperature. The magnitude of the average A-site cationic radius $\langle r_{\rm A} \rangle$ of th average A-site cationic radius $\langle r_A \rangle$ of the perovskite ABO₃ may play a crucial role. As a vacancy must have an Mn. The theoretical M_{sh} value for full alignment of the average radius $\langle r_{\text{v}} \rangle \neq 0$, the T_{c} decrease due to the Mn ion spins (using $S = 4/2$ for Mn³⁺ and $S = 3/2$ for strontium deficiency in the lacun Mn. The experimental value is smaller than the calculated [29] by a vacancy average radius $\langle r_v \rangle$ smaller than Sr²⁺ one which may indicate a canted spin state at low (1.31 Å). Such behavior has been observed in stront one which may indicate a canted spin state at low (1.31 Å) . Such behavior has been observed in strontium temperature. The angle ϕ between the average magnetic deficient Pr_{α} , Sr_{α} , MnO₂ [25]. Finally the pr deficient $Pr_{0.7}Sr_{0.3-x-x}MnO_3$ [25]. Finally the presence of moments of the Mn cations in our sample can be obtained vacancies traps the carriers and then lowers the double

At zero applied field, resistivity measurements (Fig. 7) shows a semiconductor-metallic-like transition with decreasing temperature. The maximum of the resistivity is The angle ϕ is found to be 64°. The temperature depen-
obtained at $T_o = 210$ K. This value is far below the Curie temperature, T_c =360 K. The strontium deficiency leads to an increase of the resistivity values and a net decrease of the electrical transition temperature.

> In the ferromagnetic range, the resistivity does not drop to very small values as in good metallic samples. In fact we deal with a granular material, thus there is a possibility that more or less insulating barriers develop at the grains boundaries. These barriers will limit the residual resistivity. This mechanism is well known in ceramics, in ferrite as well as in high T_c superconductors where insulating barriers cause the appearance of Josephson junctions below the transition temperature of the super-conducting grains [30].

> As illustrated in Fig. 8, the effect of the magnetic field is to lower the value of the resistivity and to weaken its temperature dependence. Meanwhile the resistivity maxi-

Fig. 7. Temperature dependence of the resistivity for both $La_{0.67}Sr_{0.15}\square_{0.18} MnO_3$ $La_{0.67}Sr_{0.15}\Box_{0.18}MnO_3$ (\bullet) and $La_{0.67}Sr_{0.33}MnO_3$ (\bullet) samples.

$$
\frac{\rho(H,T) - \rho(0,T)}{\rho(0,T)} \times 100
$$

Fig. 9. Evolution of Ln (ρ) as function of $1/T$ for

thermally activated indicating polaron conduction. Above
 T_{ρ} , the zero field resistivity data fit quite well the formulae
 $\rho = \rho_0 \exp(E_a/kT)$ where E_a is the activation energy (Fig.

activation energy (Fig.

activati $\rho = \rho_0 \exp(E_a/kT)$ where E_a is the activation energy (Fig.

9). At zero applied field, the activated energy is found to

be 0.35 eV. On applying a magnetic field *H*, the activation

energy decreases with increasing *H*, rufus attention of the technical magnetization is due to the alignment of magnetization in neighboring grains, which $\frac{1}{2}$ occurs below 3 kOe in the ferromagnetic range. This MR is maximum at low temperature where $\rho(0,T)$ is minimum.
The amplitude of this term is about 15% of the total
the total resistivity at 20 K. The second regime situated above 3 kOe corresponds to a slower decrease of the resistivity, with a larger slope near the electrical transition temperature $T_o = 210$ K. This behavior is similar to the spin– disorder resistivity of ferromagnetic materials [31], but with a carrier density modified by the semiconducting– metal transition. These ferromagnetic grains being separated by high resistivity grain boundaries. Our sample exhibits a magnetoresistance behavior at low temperature of about 30% for a magnetic applied field of about 4 T.

4. Conclusion

We have investigated the structural, magnetic and magneto-transport properties of a deficient $La_{0.67}Sr_{0.15}\Box_{0.18}MnO_3$ powder sample. Our sample crystallizes in the rhombohedral structure.

Magnetic measurements show a paramagnetic–fer-Fig. 8. Temperature dependence of the resistivity for romagnetic transition at $T_c = 360$ K. The strontium de- $La_{0.67}Sr_{0.15}\Box_{0.18}MnO_3$ at different applied magnetic fields. ficiency leads to a small decrease of T_c which can be

Fig. 10. Isotherms of MR for $La_{0.67}Sr_{0.15}\Box_{0.18}MnO_3$ from 20 to 300 K. (1995) 5144.

explained both by an increase of the Mn^{4+} content, a
decrease of the ionic redius $\langle r \rangle$ of the A cite and pinning [24] L. Laroussi, C. Boudaya, E. Dhahri, J.C. Joubert, A. Cheikhdecrease of the ionic radius $\langle r_A \rangle$ of the A site and pinning
of charge carriers by vacancies. Electrical investigations
of charge carriers by vacancies. Electrical investigations
[25] W. Boujelben, A. Cheikh-Rouhou, M. show a semiconducting–metallic like transition at 210 K Stat. Sol. (a) 181 (2000) 451. related to the granular nature of the material. Our sample [26] W. Boujelben, A. Cheikh-Rouhou, J. Pierre, J.C. Joubert, J. Alloys exhibits a magnetoresistance effect at low temperature of $\qquad \qquad \text{Comp. (2000) in press.}$
about 30% for a magnetic applied field of about 4 T [27] W. Boujelben, A. Cheikh-Rouhou, J. Pierre, J.C. Joubert, J. Magn. about 30% for a magnetic applied field of about 4 T. $\frac{[27]}{2}$ W. Boujelben, A. Cheikh-Roughlen, A. Cheikh-Roughlen, A. Cheikh-Roughlen, Magn. Mat. (2001) in press.

- Phys. Rev. Lett. 71 (1993) 2331.
- [2] R.D. Sanchez, J. Rivas, C.V.Vazquez, A.L. Quintela, M.T. Causa, M. Tovar, S. Oseroff, Appl. Phys. Lett. 68 (1996) 134.
- [3] W. Zhang, I.W. Boyd, N.S. Cohen, Q.T. Quentin, A. Pankhaurst, Appl. Surf. Sci. 109 (1997) 350.
- [4] F. Damay, C. Martin, M. Hervieu, A. Maignan, B. Raveau, G. André, F. Bourée, J. Magn. Magn. Mater. 184 (1998) 71.
- [5] A. Peles, H.P. Kunkel, X.Z. Zhou, G. Williams, J. Phys.: Condens. Mat. 11 (1999) 8111.
- [6] C. Martin, A. Maignan, M. Hervieu, B. Raveau, Phys. Rev. B 60 (1999) 12191.
- [7] W. Boujelben, A. Cheikh-Rouhou, M. Ellouze, J.C. Joubert, Phys. Stat. Sol. (a) 177 (2000) 503.
- [8] W. Boujelben, A. Cheikh-Rouhou, M. Ellouze, J.C. Joubert, Phase Transition 71 (2000) 127.
- [9] J. Heremans, J. Phys. D 26 (1993) 1149.
- [10] S. Jin, M. McCormack, T.H. Tiefel, R. Ramesh, J. Appl. Phys. 76 (1994) 6929.
- [11] K. Derbyshire, E. Korezynski, Solid State Technol. (Sept.) (1995) 57.
- [12] C. Zener, Phys. Rev. 81 (1951) 440.
- [13] P.G. De Gennes, Phys. Rev. 118 (1960) 141.
- [14] R. Mahesh, R. Mahendiran, A.K. Raychaudhuri, C.N.R. Rao, J. Solid State Chem. 114 (1995) 297.
- [15] F. Damay, C. Martin, A. Martin, B. Raveau, J. Appl. Phys. 81 (1997) 1372.
- [16] N. Abdelmoula, E. Dhahri, K. Guidara, J.C. Joubert, Phase Transitions 69 (1999) 215.
- [17] L.M. Rodriguez-Martinez, J.P. Attfield, Phys. Rev. B 54 (1996) 15622.
- [18] F. Damay, C. Martin, A. Maignan, B. Raveau, J. Appl. Phys. 82 (1997) 6181.
- [19] A.K.M. Akther Hossain, L.F. Cohen, T. Kodenkandeth, J. Mac-Manus-Driscoll, N. McNalford, J. Magn. Magn. Mater. 195 (1999) 31.
- [20] I.O. Troyanchuk, S.V. Trukhanov, H. Szymezak, K. Baerner, J. Phys.: Condens. Mat. 12 (2000) L155.
- [21] N. Abdelmoula, K. Guidara, A. Cheikh-Rouhou, E. Dhahri, J.C. Joubert, J. Solid State Chem. 151 (2000) 139.
- [22] A.J. Millis, P.B. Littlewood, B.I. Shraiman, Phys. Rev. Lett. 74
- [23] L. Laroussi, J.C. Joubert, E. Dhahri, J. Pierre, A. Cheikh-Rouhou,
-
-
-
-
- [28] G.H. Jonker, Physica 22 (1956) 707.
- [29] H.Y. Hwang, S.W. Cheong, P.G. Radaelli, M. Marezio, B. Batlogg, **References** Phys. Rev. Lett. 75 (1995) 914.
	- [30] T. Grenet, Thesis, Joseph Fourrier University, Grenoble, 1992.
- [31] P.G. de Gennes, J. Friedel, J. Phys. Chem. Solids 4 (1958) 71. [1] R. Von Helmolt, J. Wecker, B. Holzapfel, L. Schutz, K. Samwer,